Article

General and Efficient Insertion of Carbons Carrying **Benzotriazole**

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Anions formed from the lithiation of 1-(1-benzotriazolylalkyl)benzotriazoles (1, 6) and 1-(1methylthioalkyl)benzotriazoles (10 and 10a) with *n*-BuLi underwent additions to cyclic and acyclic ketones giving intermediates **3a**-**f**, **7b**-**f**, and **11b**-**d**, respectively, in excellent yields. Thermal rearrangements of intermediates **3a**,**b**,**d**–**f** and **7b**–**d**,**f** in the presence of zinc bromide provided one-carbon chain-extended or ring-expanded α -benzotriazolyl ketones **4a**,**b**,**d**-**f** and **8b**-**d**,**f** in moderate yields with excellent regioselectivity. By contrast, intermediates **11b**-**d** on treatment with zinc bromide loose a molecule of benzotriazole followed by intramolecular cyclization of the resulting intermediates **12b**-**d** to provide the 2,3- and 1,2,3-substituted indenes **13b**-**d** in good yields.

Introduction

The homologation of aldehydes and ketones is an important transformation in modern synthetic organic chemistry that has been extensively investigated. Published procedures for the insertions of a single carbon next to a carbonyl group have been summarized in several reviews¹ and in our recent publications.² These published results (Scheme 1) can be classified as follows: (i) insertion by diazo compounds;^{1d-f,3} (ii) insertion via β -oxido carbenoid intermediates generated from dibromomethane⁴ or dichloromethane;⁵ (iii) insertions utilizing α -lithioalkyl sulfoxide,⁶ selenoxide,^{6,7} α -lithioalkyl sulfones,8 or selenones;9 (iv) rearrangement of a 2-benzotriazolylalkoxide.^{2,10}

SCHEME 1



Recent work in our group achieved good regioselectivity in the simultaneous homologation and α -functionalization of ketones and aldehydes via benzotriazole intermediates.² In particular, various alkyl, aryl, heteroaryl, alkoxy, and aryloxy substituents were thus introduced.

The present manuscript deals with the insertions of carbon carrying a benzotriazolyl substituent to give α -benzotriazolyl ketones as products. Such ketones with an α -benzotriazolyl group are of significant synthetic utility for transformations to (i) diketones,¹¹ (ii) olefins,¹² (iii) for directed regioselective α -alkylation,¹³ and (iv) heterocycle ring synthesis (Scheme 2).¹⁴

Our present investigations have led to regioselective α -carbon insertions into ketones **2a**-f (Schemes 3 and

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SCHEME 2



SCHEME 3^a



^{*a*} For designation of R^1-R^2 see Table 1.

SCHEME 4^a



^a For designation of R¹-R² see Table 2.

4) utilizing 1-(1-benzotriazolylalkyl)benzotriazoles 1 and 6. In related work, the use of 1-(1-methylthioalkyl)-benzotriazoles 10 and 10a led to the 2,3- and 1,2,3-substituted indenes 13b-d.

Results and Discussion

We have examined the behavior of 1-(1-benzotriazolylalkyl)benzotriazoles 1 and 6 and 1-(1-methylthioalkyl)benzotriazoles 10 and 10a in the one-carbon homologation of ketones 2a-f and found that reactivity of intermediates 3a-f, 7b-f, and 11b-d varies with the degree and type of substitution at the α -carbon adjacent to the benzotriazole group.

The treatment of compound **1** with *n*-BuLi (1 equiv) at -78 °C under nitrogen in THF for 1 h followed by reactions with ketones **2a**-**f** (1 equiv) at -78 °C for 1 h led to high isolated yields of the corresponding intermediates **3a**-**f** (Scheme 3, Table 1). Surprisingly, under the same reaction conditions, the lithium derivative of **6**

TABLE 1. Preparation of Intermediates 3 and α -Benzotriazolyl Ketones 4

	2		yield, %/solvent/time, h	
	R ¹	R ²	3	4
a	Bn	Bn	60/THF/2	54/CHCl ₂ CHCl ₂ /5
b	Me	Ph	90/THF/2	75/CHCl ₂ CHCl ₂ /5
c	Ph	Ph	85/THF/2	-
d	-(C	H ₂) ₄ –	60/THF/2	70/CHCl ₂ CHCl ₂ /5
e	–(C	H ₂) ₅ —	80/THF/2	41/CHCl ₂ CHCl ₂ /5
f	Me	<i>i</i> -Pr	90/THF/2	60/CHCl ₂ CHCl ₂ /5

TABLE 2. Preparation of Intermediates 7 and $\alpha\text{-Benzotriazolyl}$ Ketones 8

		2	yield, 9	yield, %/solvent/time, h		
	R ¹	R ²	7	8		
b	Me	Ph	90/THF/13	40/CHCl2CHCl2/0.25		
с	Ph	Ph	70/THF/13	55/CH ₂ ClCH ₂ Cl/0.5		
d	-(C	$H_2)_4 -$	50/THF/13	35/CHCl ₂ CHCl ₂ /0.25		
е	-(C	$H_{2})_{5}-$	75/THF/13	а		
f	Me	<i>i</i> -Pr	80/THF/13	20/CH ₂ ClCH ₂ Cl/4		
^a Product 9 was isolated in 40% yield.						

reacted sluggishly and apparently reversibly with ketones **2b**-**f**: if the temperature of the reaction mixture rises to 20 °C prior to quenching with aqueous ammonium chloride, the starting compound **6** is recovered. However, keeping the reaction mixture at -78 °C for 12 h followed by quenching with aqueous ammonium chloride at the same temperature gave **7b**-**f** in high yields (Scheme 4, Table 2). The structures of compounds **3a**-**f** and **7b**-**f** were supported by their ¹H NMR and ¹³C NMR spectra (see the Experimental Section).

The mechanism of transformation of 3 to 4 and 7 to 8 was explained in our previous publications.^{2a,c,d} The present rearrangements proceed regioselectively: treating adducts 3a,b,d-f and 7b-d,f in 1,1,2,2-tetrachloroethane or 1,2-dichloroethane under reflux with a 5-fold excess of zinc bromide afforded the one carbon homologated ketones 4a,b,d-f and 8b-d,f respectively (Schemes 3 and 4, Tables 1 and 2). In each case, migration occurs of the carbon that can best stabilize an electron deficiency in the transition state. In the case of **4a**,**b**,**d**-**f**, the rearrangement could be completed in refluxing 1,1,2,2tetrachloroethane for 5 h (Table 1). By contrast, adduct 3c (Scheme 3) failed to undergo the zinc bromidepromoted rearrangement in the 1,1,2,2-tetrachloroethane under reflux for 64 h; under these reaction conditions (Scheme 3), only the dehydration product 5 was isolated in 20% yield.

We found that the rearrangements of the intermediates **7b**-**d**,**f** (Scheme 4, Table 2) required different reaction temperatures and reaction time to complete the formation of the corresponding ketones **8b**-**d**,**f**. Moreover, rearrangement of **7b**-**d**,**f** generally gave yields of **8b**-**d**,**f** around 20% less as compared to rearrangement of **3a**,**b**,**d**-**f** into **4a**,**b**,**d**-**f** probably due to the concurrent processes of dehydration and benzotriazole elimination, as in the case of the transformation of **7e** to **9** (Scheme 4). The structures of compounds **4a**,**b**,**d**-**f**, **5**, **8b**-**d**,**f**, and **9** were supported by their ¹H NMR and ¹³C NMR spectra (see the Experimental Section).

We have previously demonstrated that 1-[(methylthio)methyl]benzotriazole undergoes deprotonation with bu-

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 TABLE 3.
 Preparation of Intermediates 11b-d and 12c

 and Substituted Indenes 13b-d

			yield, %/solvent/time, h				
	R	11	12	13			
b c d	Me Ph Ph	81/THF/2 98/THF/2 96/THF/2	a 89/THF/0.25 a	30/CH ₂ ClCH ₂ Cl/0.1 70/CH ₂ ClCH ₂ Cl/0.1 78/CH ₂ ClCH ₂ Cl/0.1			
^a Intermediate was not isolated.							

SCHEME 5



tyllithium and that carbanion reacts readily with aliphatic and aromatic ketones followed by rearrangement of the resulting intermediates under zinc bromide catalysis to afford the one-carbon homologated α -(methylthio)methyl ketones.^{2c} We now show that the 1-(1methylthioalkyl)benzotriazoles 10 and 10a similarly undergo smooth deprotonation with *n*-BuLi (1 equiv) at -78 °C under nitrogen in THF for 1 h and the resulting carbanions react cleanly with ketones **2b**,**c** (1 equiv) at -78 °C for 1 h giving excellent isolated yields of adducts (Schemes 5 and 6, Table 3). However, the adducts 11b-d failed to undergo rearrangement to the α -carbon insertion products, instead the elimination of benzotriazole was observed and the 2,3- and 1,2,3-substituted indenes 13b-d were isolated (Schemes 5 and 6, Table 3). In fact, the adducts 11b-d react with 3-fold excess of zinc bromide for 15 min in 1,2-dichloromethane at 20 °C loosing a molecule of benzotriazole providing intermediates **12b**-**d** which, under reflux in the same solvent for 5 min undergo intramolecular cyclization to give 13b-d in good yields (Schemes 5 and 6, Table 3). Formation of 12b-d has been established by the isolation of 12c in 89% yield, while 12b was hydrolyzed after aqueous workup to give 14 in 75% yield.

In preceding papers,^{2a,c,d} it was shown that monosubstituted benzotriazolylmethanes such as 1-(arylmethyl)-, 1-(heteroarylmethyl)-, 1-(alkenylmethyl)-, 1-(alkoxymethyl)-, and 1-[(phenylthio)methyl]benzotriazoles are excellent insertion reagents for the preparation of α -functionalized ketones. We have now found that (1*H*-benzotriazol-1-ylmethyl)-1*H*-benzotriazole **1** also behaves in the same manner undergoing α -carbon insertions into ketones 2, while in the cases of disubstituted benzotriazolyl methane derivatives 6, 10, and 10a, either poorer yields of inserted ketones 8b-d, f were observed or a different reaction pathway was involved leading to the formation of indenes 13b-d.

In summary, an efficient method for the α -carbon insertions of 1-(1-benzotriazolylalkyl)benzotriazoles into ketones has been demonstrated. The insertion reactions take place regioselectively with wide set of cyclic and acyclic ketones providing both one-carbon chain-extended and ring-expanded α -benzotriazolyl ketones. In certain cases, 1-(1-methylthioalkyl)benzotriazoles have been utilized as substrate for the synthesis of 2,3- and 1,2,3-substituted indenes.

Experimental Section

General Methods. Melting points were determined on a hot-stage apparatus and are uncorrected. NMR spectra were recorded in CDCl₃, acetone- d_6 , or DMSO- d_6 with TMS as the internal standard for ¹H (300 MHz) or a solvent as the internal standard for ¹³C (75 MHz). THF was dried over sodium/ benzophenone and used freshly distilled. Column chromatography was conducted on silica gel 200–425 meshes. (1*H*-Benzotriazol-1-ylmethyl)-1*H*-benzotriazole **1**,¹⁵ (1*H*-benzotriazol-1-ylmethyl)-1*H*-benzotriazole **1**,¹⁶ and 1-(methylthio)-1-ethyl-1*H*-benzotriazole **10**¹⁶ were prepared according to previously reported procedures.

General Procedure for the Preparation of Intermediates 3a–f. A solution of dibenzotriazol-1-ylmethane **1** (8.00 mmol) in THF (50 mL) was cooled to -78 °C, and a solution of *n*-BuLi (8.80 mmol, 1.58 M in hexane, 5.6 mL) was added dropwise. The reaction mixture was stirred at the same temperature for 1 h, and a solution of an appropriate ketone (8.80 mmol) in THF (15 mL) was added. The mixture was stirred for an additional 1 h at -78 °C. Then water was added (30 mL), and the reaction mixture was extracted with diethyl ether. The extract was dried over anhydrous magnesium sulfate and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel to give **3a–f**.

1,1-Di(1,2,3-benzotriazol-1-yl)-2-benzyl-3-phenyl-2-propanol (3a): microcrystals from methanol (60%); mp 176–177 °C; ¹H NMR (DMSO- d_6) δ 8.11 (d, J = 8.1 Hz, 2H), 7.56–7.39 (m, 6H), 7.34 (s, 1H), 7.18–7.03 (m, 10H), 6.14 (s, 1H), 3.57 (d, J = 14.0 Hz, 2H), 3.16 (d, J = 14.0 Hz, 2H); ¹³C NMR (DMSO- d_6) δ 144.8, 135.9, 132.9, 130.4, 128.5, 127.8, 126.4, 124.7, 119.5, 110.4, 77.8, 75.2, 42.7. Anal. Calcd for C₂₈H₂₄N₆O: C, 73.02; H, 5.25; N, 18.25. Found: C, 73.26; H, 5.32; N, 18.45.

(±)-1,1-Di(1,2,3-benzotriazol-1-yl)-2-phenyl-2-propanol (3b): microcrystals from methanol (90%); mp 181–182 °C; ¹H NMR (DMSO- d_6) δ 8.46 (s, 1H), 8.38 (d, J = 8.5 Hz, 1H), 8.30 (d, J = 8.5 Hz, 1H), 8.10 (d, J = 8.3 Hz, 1H), 7.93 (d, J = 8.2 Hz, 1H), 7.80–7.68 (m, 2H), 7.68–7.50 (m, 2H), 7.50–7.27 (m, 2H), 7.27–7.04 (m, 3H), 6.49 (s, 1H), 1.81 (s, 3H); ¹³C NMR (DMSO- d_6) δ 144.8, 144.5, 143.7, 133.1, 132.9, 128.0, 127.7, 127.6, 127.1, 125.5, 124.5, 124.3, 119.1, 118.8, 112.7, 112.6, 77.2, 77.1, 27.5 Anal. Calcd for C₂₁H₁₈N₆O: C, 68.09; H, 4.90; N, 22.69. Found: C, 68.45; H, 5.10; N, 22.56.

SCHEME 6



2,2-Di(1,2,3-benzotriazol-1-yl)-1,1-diphenyl-1-ethanol (**3c**): microcrystals from toluene (85%); mp 231–232 °C; ¹H NMR (DMSO- d_6) δ 9.30 (s, 1H), 8.04 (d, J = 8.5 Hz, 2H), 7.98 (d, J = 8.3 Hz, 2H), 7.70 (d, J = 7.8 Hz, 4H), 7.50 (t, J = 7.1 Hz, 2H), 7.33 (t, J = 7.2 Hz, 2H), 7.25–7.10 (m, 6H), 7.01 (s, 1H); ¹³C NMR (DMSO- d_6) δ 144.6, 143.2, 132.6, 128.1, 127.9, 127.3, 125.8, 124.6, 119.2, 111.9, 81.3, 75.5. The spectral data of this compound is identical to that reported in the literature.¹⁸

1-[Di(1,2,3-benzotriazol-1-yl)methyl]cyclopentanol (3d): microcrystals from methanol (60%); mp 145–146 °C; ¹H NMR (CDCl₃) δ 8.05–8.01 (m, 2H), 7.54–7.49 (m, 3H), 7.40– 7.26 (m, 4H), 4.64 (s, 1H), 2.10–1.75 (m, 8H); ¹³C NMR (CDCl₃) δ 145.8, 132.3, 128.7, 124.8, 120.0, 110.9, 85.4, 77.7, 38.7, 24.0. Anal. Calcd for C₁₈H₁₈N₆O: C, 64.66; H, 5.43; N, 25.13. Found: C, 65.03; H, 5.62; N, 25.49.

1-[Di(1,2,3-benzotriazol-1-yl)methyl]cyclohexanol (**3e**): microcrystals from acetone (83%); mp 191–192 °C; ¹H NMR (CDCl₃) δ 8.03 (d, J = 8.2 Hz, 2H), 7.76 (d, J = 8.4 Hz, 2H), 7.57 (s, 1H), 7.46–7.40 (m, 2H), 7.36–7.31 (m, 2H), 4.64 (s, 1H), 1.86–1.52 (m, 9H), 1.42–1.24 (m,1H); ¹³C NMR (CDCl₃) δ 145.7, 132.5 128.7, 124.8, 120.0, 111.2, 77.4, 75.6, 34.8, 25.2, 21.2. Anal. Calcd for C₁₉H₂₀N₆O: C, 65.50; H, 5.79; N, 24.12. Found: C, 65.58; H, 5.69; N, 24.39.

(±)-1,1-Di(1,2,3-benzotriazol-1-yl)-2,3-dimethyl-2-butanol (3f): microcrystals from methanol (90%); mp 178–179 °C; ¹H NMR (CDCl₃) δ 8.06–7.99 (m, 2H), 7.85 (d, J= 8.5 Hz, 1H), 7.78 (d, J= 8.3 Hz, 1H), 7.75 (s, 1H), 7.50–7.26 (m, 4H), 4.77 (s,1H), 2.00–1.82 (m, 1H), 1.20 (s, 3H), 1.10 (d, J= 6.7 Hz, 3H), 1.04 (d, J= 6.8 Hz, 3H); ¹³C NMR (CDCl₃) δ 146.1, 145.4, 132.6, 132.3, 129.0, 128.6, 125.0, 124.7, 120.1, 119.9, 111.9, 110.7, 79.6, 75.6, 34.8, 18.2, 18.0, 16.5. Anal. Calcd for C₁₈H₂₀N₆O: C, 64.27; H, 5.99; N, 24.98. Found: C, 64.72; H, 6.02; N, 25.43.

General Procedure for the Insertion into Ketones: Preparation of 4a,b,d-f. To a solution of an appropriate intermediate **3a,b,d-f** (3 mmol) in 1,1,2,2-tetrachloroethane (20 mL) under nitrogen was added zinc bromide (15 mmol), and the reaction mixture was refluxed for 5 h. The reaction mixture was poured into the 1 N aqueous hydrochloric acid and extracted by chloroform. The extract was washed with water, dried over potassium carbonate and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel to give **4a,b,d-f**.

(±)-3-(1,2,3-Benzotriazol-1-yl)-1,4-diphenyl-2-butanone (4a): microcrystals from acetone (54%); mp 106–107 °C; ¹H NMR (CDCl₃) δ 8.06–7.98 (m, 1H), 7.36–7.26 (m, 2H), 7.18–7.13 (m, 3H), 7.10–7.04 (m, 4H), 6.90–6.85 (m, 4H), 5.67 (dd, J = 9.8, 5.6 Hz, 1H), 3.72 (dd, J = 14.2, 5.6 Hz, 1H), 3.65 (d, J = 15.6 Hz, 1H), 3.55 (d, J = 15.6 Hz, 1H), 3.50 (dd, J = 14.2, 9.8 Hz, 1H); ¹³C NMR (CDCl₃) δ 201.6, 145.9, 135.8, 133.0, 132.2, 129.1, 128.7, 128.6, 128.5, 127.7, 127.3, 127.0, 124.1, 120.0, 109.3, 67.5, 47.0, 36.1. Anal. Calcd for C₂₂H₁₉-N₃O: C, 77.40; H, 5.61; N, 12.31. Found: C, 77.06; H, 5.91; N, 12.35.

(±)-1-(1,2,3-Benzotriazol-1-yl)-1-phenylacetone (4b): microcrystals from ether (75%); mp 115–116 °C; ¹H NMR (CDCl₃) δ 8.06 (d, J = 7.3 Hz, 1H), 7.42–7.30 (m, 7H), 7.24–7.21 (m, 1H), 6.75 (s, 1H), 2.34 (s, 3H); ¹³C NMR (CDCl₃) δ 200.5, 146.2, 132.9, 132.2, 129.5, 129.2, 128.9, 127.7, 124.0, 120.1, 110.4, 71.9, 28.1. Anal. Calcd for C₁₅H₁₃N₃O: C, 71.70; H, 5.21; N, 16.72. Found: C, 72.05; H, 5.13; N, 16.87.

(±)-2-(1,2,3-Benzotriazol-1-yl)cyclohexanone (4d): microcrystals from methanol (70%); mp 129–130 °C; 1 H NMR

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 $\begin{array}{l} (CDCl_3) \ \delta \ 8.07 \ (d, \ J=8.0 \ Hz, \ 1H), \ 7.47-7.32 \ (m, \ 3H), \ 5.58 \\ (dd, \ J=12.6, \ 6.3 \ Hz, \ 1H), \ 2.76-2.53 \ (m, \ 4H), \ 2.28-2.16 \ (m, \ 2H), \ 2.04-1.78 \ (m, \ 2H); \ ^{13}C \ NMR \ (CDCl_3) \ \delta \ 202.7, \ 146.1, \ 132.9, \ 127.2, \ 123.8, \ 120.1, \ 110.3, \ 66.6, \ 41.1, \ 32.9, \ 26.8, \ 24.5. \ Anal. \\ Calcd \ for \ C_{12}H_{13}N_3O: \ C, \ 66.96; \ H, \ 6.09; \ N, \ 19.52. \ Found: \ C, \ 67.13; \ H, \ 6.17; \ N, \ 19.58. \ The spectral \ data \ of \ this \ compound \ is \ identical \ to \ that \ reported \ in \ the \ literature. \ ^{13} \end{array}$

(±)-2-(1,2,3-Benzotriazol-1-yl)cycloheptanone (4e): microcrystals from methanol (41%); mp 103–104 °C; ¹H NMR (CDCl₃) δ 8.05 (d, J = 8.3 Hz, 1H), 7.50–7.40 (m, 2H), 7.40–7.30 (m, 1H), 5.73 (dd, J = 10.3, 3.4 Hz, 1H), 2.85–2.35 (m, 4H), 2.25–1.70 (m, 5H), 1.65–1.50 (m, 1H); ¹³C NMR (CDCl₃) δ 205.8, 146.1, 132.9, 127.3, 123.8, 120.0, 110.4, 68.1, 41.5, 30.7, 29.3, 27.9, 23.6. Anal. Calcd for C₁₃H₁₅N₃O: C, 68.10; H, 6.59; N, 18.33. Found: C, 68.44; H, 6.61; N, 18.69.

(±)-3-(1,2,3-Benzotriazol-1-yl)-4-methyl-2-pentanone (4f): oil (60%); ¹H NMR (CDCl₃) δ 8.10 (d, J = 8.2 Hz, 1H), 7.57– 7.48 (m, 2H), 7.43–7.38 (m, 1H), 5.12 (d, J = 9.5 Hz, 1H), 3.20–2.98 (m, 1H), 2.10 (s, 3H), 1.12 (d, J = 6.7 Hz, 3H), 0.80 (d, J = 6.7 Hz, 3H); ¹³C NMR (CDCl₃) δ 202.5, 146.0, 132.9, 127.9, 124.3, 120.1, 109.9, 73.9, 28.5, 27.5, 19.8, 18.8. Anal. Calcd for C₁₂H₁₅N₃O: C, 66.34; H, 6.96; N, 19.34. Found: C, 65.94; H, 7.14; N, 19.70.

1-[1-(1,2,3-Benzotriazol-1-yl)-2,2-diphenylvinyl]-1,2,3benzotriazole (5). To a solution of 2,2-di(1,2,3-benzotriazol-1-yl)-1,1-diphenyl-1-ethanol 3c (1 g, 2.3 mmol) in 1,1,2,2tetrachloroethane (20 mL) under nitrogen was added zinc bromide (3.4 g, 11.6 mmol), and the reaction mixture was refluxed for 64 h. The reaction mixture was poured into the 1 N aqueous hydrochloric acid and extracted by chloroform. The chloroform solution was washed with water, dried over potassium carbonate, and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel to give the pure product **5** as microcrystals (20%): mp 229– 230 °C; ¹H NMR (CDCl₃) δ 8.00 (d, J = 8.0 Hz, 2H), 7.40– 7.27 (m, 6H), 7.17–7.10 (m, 10H); 13 C NMR (CDCl₃) δ 145.3, 142.1, 136.9, 132.7, 129.2, 129.1, 128.8, 128.5, 124.6, 122.0, 120.2, 109.7. Anal. Calcd for C₂₆H₁₈N₆: C, 75.35; H, 4.38; N, 20.28. Found: C, 75.22; H, 4.17; N, 20.40.

General Procedure for the Preparation of Intermediates 7b–f. A solution of 1-[1-(1,2,3-benzotriazol-1-yl)ethyl]-1,2,3-benzotriazole **6** (7.6 mmol) in THF (50 mL) was cooled to -78 °C, and a solution of *n*-BuLi (8.6 mmol, 1.58 M in hexane, 5.5 mL) was added dropwise. The reaction mixture was stirred at the same temperature for 1 h, and a solution of an appropriate ketone (8.3 mmol) in THF (15 mL) was added. The mixture was stirred for an additional 12 h at -78 °C. Then water was added (30 mL), and the reaction mixture was extracted with ether. The extract was dried over anhydrous magnesium sulfate and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel to give **7b–f**.

(±)-3,3-Di(1,2,3-benzotriazol-1-yl)-2-phenyl-2-butanol (7b): microcrystals from methanol (90%); mp 211–212 °C; ¹H NMR (CDCl₃) δ 8.10 (d, J = 8.4 Hz, 1H), 8.00 (d, J = 8.2 Hz, 1H), 7.30–6.85 (m, 9H), 6.28 (d, J = 8.4 Hz, 1H), 6.04 (br s, 1H), 5.61 (s, 1H), 2.62 (s, 3H), 2.33 (s, 3H); ¹³C NMR (CDCl₃) δ 145.8, 1 45.7, 140.6, 133.0, 132.6, 128.5, 128.1, 127.6, 127.6, 126.4, 124.7, 124.3, 120.5, 119.9, 111.2, 111.2, 87.4, 81.7, 26.5, 24.2. Anal. Calcd for C₂₂H₂₀N₆O: C, 68.73; H, 5.24; N, 21.86. Found: C, 68.59; H, 5.36; N, 22.13.

2,2-Di(1,2,3-benzotriazol-1-yl)-1,1-diphenyl-1-propanol (7c): microcrystals from methanol (70%); mp 221–222 °C; ¹H NMR (CDCl₃) δ 8.01 (d, J = 8.4 Hz, 2H), 7.30–7.00 (m, 14H), 6.42 (d, J = 6.6 Hz, 2H), 5.68 (s, 1H), 3.14 (s, 3H); ¹³C NMR (CDCl₃) δ 146.0, 143.0, 133.1, 129.4, 128.3, 127.9, 127.3, 124.4, 120.2, 111.4, 87.5, 87.1, 24.6. Anal. Calcd for C₂₇H₂₂-N₆O: C, 72.63; H, 4.97; N, 18.82. Found: C, 72.52; H, 5.14; N, 18.76.

1-[1,1-Di(1,2,3-benzotriazol-1-yl)ethyl]cyclopentanol (7d): microcrystals from methanol (50%); mp 223–224 °C; ¹H NMR (DMSO- d_6) δ 8.08 (d, J = 8.3 Hz, 2H), 7.29–7.23 (m,

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2H), 7.15–7.10 (m, 2H), 6.24 (d, J = 8.4 Hz, 2H), 5.35 (s, 1H), 2.62 (s, 3H), 2.49–2.33 (m, 2H), 1.90–1.60 (m, 6H); ¹³C NMR (DMSO- d_6) δ 145.3, 132.4, 127.6, 124.0, 119.4, 111.5, 87.4, 86.8, 37.2, 23.7, 23.6. Anal. Calcd for C₁₉H₂₀N₆O: C, 65.50; H, 5.79; N, 24.12. Found: C, 65.90; H, 6.03; N, 24.60.

1-[1,1-Di(1,2,3-benzotriazol-1-yl)ethyl]cyclohexanol (7e): microcrystals from methanol (75%); mp 249–250 °C; ¹H NMR (DMSO- d_6) δ 8.07 (d, J = 8.2 Hz, 2H), 7.28–7.23 (m, 2H), 7.16–7.10 (m, 2H), 6.26 (d, J = 8.4 Hz, 2H), 5.34 (s, 1H), 2.60 (s, 3H), 2.10 (br s, 2H), 1.85–1.48 (m, 7H), 1.35–1.05 (m, 1H); ¹³C NMR (DMSO- d_6) δ 145.2, 132.7, 127.9, 124.3, 119.6, 111.9, 88.2, 78.3, 32.1, 24.9, 23.6, 21.2. Anal. Calcd for $C_{20}H_{22}N_6O$: C, 66.28; H, 6.12; N, 23.19. Found: C, 66.22; H, 6.58; N, 23.32.

(±)-2,2-Di(1,2,3-benzotriazol-1-yl)-3,4-dimethyl-3-pentanol (7f): microcrystals from methanol (80%); mp 170–172 °C; ¹H NMR (DMSO- d_6) δ 8.06 (dd, J = 8.2, 2.9 Hz, 2H), 7.29– 7.22 (m, 2H), 7.17–7.09 (m, 2H), 6.45 (d, J = 8.5 Hz, 1H), 6.24 (d, J = 8.1 Hz, 1H), 5.44 (s, 1H), 2.65 (s, 3H), 2.46–2.31 (m, 1H), 1.45 (s, 3H), 0.97 (d, J = 6.4 Hz, 3H), 0.82 (d, J = 6.7 Hz, 3H); ¹³C NMR (DMSO- d_6) δ 145.3, 145.1, 132.7, 132.5, 127.9, 127.8, 124.3, 124.2, 119.5, 119.5, 112.1, 112.0, 89.0, 80.7, 33.9, 24.3, 20.4, 19.8, 18.7. Anal. Calcd for C₁₉H₂₂N₆O: C, 65.12; H, 6.33; N, 23.98. Found: C, 65.00; H, 6.53; N, 24.29.

(±)-3-(1,2,3-Benzotriazol-1-yl)-3-phenyl-2-butanone (8b). To a solution of 3,3-di(1,2,3-benzotriazol-1-yl)-2-phenyl-2-butanole 7b (0.4 g, 1 mmol) in 1,1,2,2-tetrachloroethane (20 mL) under nitrogen was added zinc bromide (1.2 g, 5.3 mmol), and the reaction mixture was refluxed for 15 min. The reaction mixture was poured into the 1 N aqueous hydrochloric acid and extracted by chloroform. The chloroform solution was washed with water, dried over potassium carbonate and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel to give 8b as an oil (40%): ¹H NMR (CDCl₃) δ 8.10 (d, J = 8.2 Hz, 1H), 7.42–7.24 (m, 7H), 6.66 (d, J = 8.2 Hz, 1H), 2.27 (s, 3H), 2.20 (s, 3H); ¹³C NMR (CDCl₃) & 202.1, 146.4, 137.0, 132.8, 129.0, 128.9, 127.3, 127.0, 123.9, 120.2, 112.0, 75.2, 26.4, 24.8. Anal. Calcd for C₁₆H₁₅N₃O: C, 72.43; H, 5.70; N, 15.84. Found: C, 72.70; H, 5.77; N, 15.96.

(±)-2-(1,2,3-Benzotriazol-1-yl)-1,2-diphenyl-1-propanone (8c). To a solution of 2,2-di(1,2,3-benzotriazol-1-yl)- $\hat{1}, 1$ -diphenyl-1-propanole **7c** (0.3 g, 0.67 mmol) in 1,2-dichlorethane (20 mL) under nitrogen was added zinc bromide (1.2 g, 3.5 mmol), and the reaction mixture was refluxed for 30 min. The reaction mixture was poured into the 1 N aqueous hydrochloric acid and extracted by chloroform. The chloroform solution was washed with water, dried over potassium carbonate and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel to give 8c as an oil (55%): ¹H NMR (CDCl₃) δ 8.06 (d, J = 8.0 Hz, 1H), 7.56– 7.53 (m, 2H), 7.43-7.20 (m, 10H), 6.74 (d, J = 8.2 Hz, 1H), 2.61 (s, 3H); ¹³C NMR (CDCl₃) δ 196.5, 146.6, 138.4, 135.3, 133.5, 132.7, 129.4, 129.0, 128.8, 128.3, 127.9, 127.3, 123.8, 120.2, 111.9, 75.1, 27.6. Anal. Calcd for C₂₁H₁₇N₃O: C, 77.04; H, 5.23; N, 12.83. Found: C, 77.00; H, 5.27; N, 12.80.

(±)-2-(1,2,3-Benzotriazol-1-yl)-2-methylcyclohexanone (8d). To a solution of 1-[1,1-di(1,2,3-benzotriazol-1-yl)ethyl]cyclopentanole 7d (0.3 g, 0.86 mmol) in 1,1,2,2-tetrachloroethane (20 mL) under nitrogen was added zinc bromide (1 g, 4.4 mmol) and the reaction mixture was refluxed for 15 min. The reaction mixture was poured into the 1 N aqueous hydrochloric acid and extracted by chloroform. The chloroform solution was washed with water, dried over potassium carbonate, and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel to give 8d as an oil (35%): ¹H NMR (CDCl₃) δ 8.10 (d, J = 8.0 Hz, 1H), 7.47– 7.30 (m, 3H), 3.52-3.34 (m, 1H), 2.57-2.49 (m, 1H), 2.36-2.25 (m, 1H), 2.16-1.83 (m, 5H), 1.73 (s, 3H); ¹³C NMR (CDCl₃) δ 207.1, 146.6, 131.8, 127.5, 123.9, 120.2, 110.6, 70.0, 39.5, 39.1, 28.0, 23.0. 21.3. Anal. Calcd for C13H15N3O: C, 68.10; H, 6.59; N, 18.33. Found: C, 67.85; H, 6.67; N, 18.50.

(±)-3-(1,2,3-Benzotriazol-1-yl)-3,4-dimethyl-2-pentanone (8f). To a solution of 2,2-di(1,2,3-benzotriazol-1-yl)-3,4-dimethyl-3-pentanole 7f (0.4 g, 1.14 mmol) in 1,2-dichlorethane (20 mL) under nitrogen was added zinc bromide (1.3 g, 5.8 mmol), and the reaction mixture was refluxed for 4 h. The reaction mixture was poured into the 1 N aqueous hydrochloric acid and extracted by chloroform. The chloroform solution was washed with water, dried over potassium carbonate and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel to give **8f** as an oil (20%): ¹H NMR (CDCl₃) δ 8.12 (d, J = 7.1 Hz, 1H), 7.46– 7.35 (m, 3H), 3.28 (q, J = 6.7 Hz, 1H), 1.98 (s, 3H), 1.86 (s, 3H), 1.09 (d, J = 7.7 Hz, 3H), 0.82 (d, J = 7.7 Hz, 3H); ¹³C NMR (CDCl₃) δ 204.9, 146.7, 132.3, 127.6, 124.1, 120.3, 111.0, 75.4, 31.3, 24.4, 18.0, 17.3, 15.8. Anal. Calcd for C₁₃H₁₇N₃O: C, 67.51; H, 7.41; N, 18.17. Found: C, 67.87; H, 7.85; N, 18.33.

1-[1-(1-Cyclohexen-1-yl)vinyl]-1,2,3-benzotriazole (9). To a solution of 1-[1,1-di(1,2,3-benzotriazol-1-yl)ethyl]cyclohexanole 7e (0.3 g, 0.83 mmol) in 1,1,2,2-tetrachloroethane (20 mL) under nitrogen was added zinc bromide (0.95 g, 4.2 mmol) and the reaction mixture was refluxed for 5 min. The reaction mixture was poured into the 1 N aqueous hydrochloric acid and extracted by chloroform. The chloroform solution was washed with water, dried over potassium carbonate, and evaporated under reduced pressure. The residue was purified by column chromatography on silica gel to give 9 as an oil (35%): ¹H NMR (CDCl₃) δ 8.08 (dd, J = 8.2, 0.8 Hz, 1H), 7.49– 7.47 (m, 2H), 7.41–7.35 (m, 1H), 5.57 (s, 1H), 5.46 (t, J = 4.1 Hz, 1H), 5.35 (s, 1H), 2.35-2.30 (m, 2H), 2.12-2.03 (m, 2H), 1.81–1.73 (m, 2H), 1.67–1.59 (m, 2H); ¹³C NMR (CDCl₃) δ 145.4, 143.7, 133.6, 131.9, 129.9, 127.7, 123.9, 119.8, 110.6, 110.2, 25.5, 25.4, 22.2, 21.6. Anal. Calcd for C14H15N3: C, 74.64; H, 6.71; N, 18.65. Found: C, 74.60; H, 7.01; N, 18.53.

(±)-1-(1,2,3-Benzotriazol-1-yl)propyl Methyl Sulfide (10a). The solution of 1-(1,2,3-benzotriazol-1-yl)methyl methyl sulfide (2 g, 11.20 mmol) in THF (60 mL) was cooled to -78°C, and the solution of *n*-BuLi (12.3 mmol, 1.58 M in hexane, 7.9 mL) was added dropwise. The reaction mixture was stirred at the same temperature for 1 h, and the solution of ethyl iodide (1.92 g, 12.3 mmol) in THF (15 mL) was added. The mixture was stirred for an additional 1 h at -78 °C. Then water was added (30 mL), and the product was extracted with ether. The residue was purified by gradient column chromatography using the mixture of hexanes/EtOAc (9:1-3:1) to give **10a** as an oil (82%): ¹H NMR (CDCl₃) δ 8.09 (d, J = 8.2 Hz, 1H), 7.83 (d, J = 8.4 Hz, 1H), 7.52–7.46 (m, 1H), 7.44–7.36 (m, 1H), 5.85 (dd, J = 8.8, 6.6 Hz, 1H), 2.50–2.26 (m, 2H), 1.89 (s, 3H), 0.94 (t, J = 7.4 Hz, 3H); ¹³C NMR (CDCl₃) δ 147.0, 131.1, 127.1, 124.1, 120.2, 111.3, 67.6, 27.6, 13.7, 11.3. Anal. Calcd for C₁₉H₂₂N₆O: C, 57.94; H, 6.32; N, 20.27. Found: C, 57.97; H, 6.36; N, 20.65.

General Procedure for the Preparation of Intermediates (11b,c). The solution of 1-(1-methylthioethyl)-1,2,3-benzotriazole **10** (1 g, 5.18 mmol) in THF (50 mL) was cooled to -78 °C, and the solution of *n*-BuLi (5.70 mmol, 1.58 M in hexane, 3.7 mL) was added dropwise. The reaction mixture was stirred at the same temperature for 1 h, and the solution of corresponding ketones (5.70 mmol) in THF (15 mL) was added. The mixture was stirred for an additional 1 h at -78 °C. Then water was added (30 mL), and the product was extracted with ether. The rescue was purified by crystalization from ether to give **11b,c**.

(±)-3-(1,2,3-Benzotriazol-1-yl)-3-(methylthio)-2-phenyl-2-butanol (mixture of two diastereoisomers) (11b): microcrystals from acetone (81%); mp 116–117 °C; ¹H NMR (CDCl₃) δ 8.35 (d, J = 8.2 Hz, 1H), 8.30 (d, J = 7.9 Hz, 1H), 7.98 (d, J = 7.6 Hz, 1H), 7.93 (d, J = 8.1 Hz, 1H), 7.50–7.20 (m, 10H), 7.20–7.00 (m, 2H), 6.97–6.85 (m, 2H), 2.98 (s, 1H), 2.88 (s, 1H), 2.27 (s, 3H), 2.26 (s, 3H), 1.92 (s, 3H), 1.72 (s, 3H), 1.66 (s, 3H), 1.53 (s, 3H); ¹³C NMR (CDCl₃) δ 146.2, 146.1, 142.7, 141.8, 134.5, 134.1, 127.8, 127.7, 127.4, 127.2, 126.6, 126.5, 126.4, 126.4, 123.7, 123.5, 119.3, 119.2, 116.0, 115.9, 81.0, 80.8, 80.3, 80.2, 26.4, 25.8, 24.7, 24.5, 11.7, 11.5. Anal. Calcd for $C_{17}H_{19}N_3OS\colon$ C, 65.15; H, 6.11; N, 13.41. Found: C, 64.49; H, 6.25; N, 12.83.

(±)-2-(1,2,3-Benzotriazol-1-yl)-2-(methylthio)-1,1-diphenyl-1-propanol (11c): microcrystals from acetone (98%); mp 139–140 °C; ¹H NMR (CDCl₃) δ 8.00–7.84 (m, 2H), 7.58–7.47 (m, 2H), 7.42–7.10 (m, 10H), 4.19 (s, 1H), 2.46 (s, 3H), 1.71 (s, 3H); ¹³C NMR (CDCl₃) δ 146.1, 143.0, 142.9, 134.0, 129.0, 128.3, 127.7, 127.5, 127.4, 127.3, 126.6, 123.6, 119.5, 115.4, 85.2, 80.6, 25.3, 12.7. Anal. Calcd for C₂₂H₂₁N₃OS: C, 70.37; H, 5.64; N, 11.19. Found: C, 70.62; H, 5.82; N, 10.95.

 $(\pm) \hbox{-} 2 \hbox{-} (1,2,3 \hbox{-} Benzotriazol \hbox{-} 1 \hbox{-} yl) \hbox{-} 2 \hbox{-} (methylthio) \hbox{-} 1,1 \hbox{-} diphen$ yl-1-butanol (11d). The solution of 1-(1,2,3-benzotriazol-1yl)propyl methyl sulfide 10a (1 g, 4.83 mmol) in THF (50 mL) was cooled to -78 °C, and the solution of *n*-BuLi (5.31 mmol, 1.58 M in hexane, 3.4 mL) was added dropwise. The reaction mixture was stirred at the same temperature for 1 h, and the solution of benzophenone (1 g, 5.31 mmol) in THF (15 mL) was added. The mixture was stirred for an additional 2 h at -78 °C. Then water was added (30 mL), and the product was extracted with ether. The rescue was purified by crystalization from acetone to give 11d as microcrystals (96%): mp 123-124 °C; ¹H NMR (CDCl₃) δ 8.22 (d, J = 8.40 Hz, 1H), 8.00-7.92 (m, 1H), 7.60-7.50 (m, 2H), 7.50-7.41 (m, 2H), 7.40-7.27 (m, 2H), 7.23-7.14 (m, 3H), 7.14-7.06 (m, 3H), 5.25 (s, 1H), 3.21-3.08 (m, 1H), 2.60-2.46 (m, 1H), 1.48 (s, 3H), 0.84 (t, J = 7.2 Hz, 3H); ¹³C NMR (CDCl₃) δ 145.4, 144.7, 142.3, 135.1, 129.5, 128.3, 127.5, 127.4, 127.1, 127.0 (2), 123.7, 119.9, 115.4, 86.2, 84.0, 29.4, 14.4, 9.5. Anal. Calcd for C23H23N3OS: C, 70.92; H, 5.95; N, 10.79. Found: C, 70.76; H, 5.98; N, 10.81.

2-(Methylthio)-1,1-diphenyl-2-propen-1-ol (12c). To a solution of 2-(1,2,3-benzotriazol-1-yl)-2-(methylthio)-1,1-diphenyl-1-propanole **11c** (0.2 g, 0.53 mmol) in 1,2-dichlorethane (10 mL) under nitrogen was added zinc bromide (0.36 g, 1.6 mmol), and the reaction mixture was stirred at 25 °C for 15 min. The solvent was evaporated under reduced pressure, and the residue was purified by column chromatography on silica gel using hexanes, to give **12c** as an oil (89%): ¹H NMR (CDCl₃) δ 7.42–7.36 (m, 4H), 7.35–7.27 (m, 6H), 5.12 (d, J = 1.1 Hz, 1H), 4.98 (d, J = 1.1 Hz, 1H), 3.29 (s, 1H), 2.29 (s, 3H); ¹³C NMR (CDCl₃) δ 153.9, 144.2, 127.8, 127.7, 127.5, 112.2, 82.7, 16.4. Anal. Calcd for C₁₆H₁₆OS: C, 74.96; H, 6.29; N, 0. Found: C, 74.68; H, 6.30; N, 0.30.

2-(Methylthio)-3-methyl-1*H***-indene (13b).** To a solution of 3-(1,2,3-benzotriazol-1-yl)-3-(methylthio)-2-phenyl-2-butanol **11b** (0.3 g, 1.60 mmol) in 1,2-dichlorethane (20 mL) under nitrogen was added zinc bromide (0.65 g, 2.87 mmol), and the reaction mixture was refluxed for 5 min. The solvent was evaporated under reduced pressure and residue was purified by column chromatography on silica gel using hexanes to give

13b as an oil (30%): ¹H NMR (CDCl₃) δ 7.37 (d, J = 7.5 Hz, 1H), 7.32–7.17 (m, 2H), 7.16–7.09 (m, 1H), 3.52–3.47 (m, 2H), 2.41 (s, 3H), 2.12 (t, J = 2.0 Hz, 3H); ¹³C NMR (CDCl₃) δ 146.3, 142.2, 136.0, 135.6, 126.5, 123.9, 122.9, 117.9, 40.7, 16.0, 11.1. Anal. Calcd for C₁₁H₁₂S: C, 74.95; H, 6.86; N, 0. Found: C, 74.94; H, 6.60; N, 0.06.

2-(Methylthio)-3-phenyl-1*H***-indene (13c).** To a solution of 2-(1,2,3-benzotriazol-1-yl)-2-(methylthio)-1,1-diphenyl-1-propanole **11c** (0.5 g, 1.33 mmol) in 1,2-dichlorethane (20 mL) under nitrogen was added zinc bromide (0.9 g, 4 mmol), and the reaction mixture was refluxed for 5 min. The solvent was evaporated under reduced pressure, and residue was purified by column chromatography on silica gel using hexanes to give **13c** as plates (70%): mp 86–87 °C; ¹H NMR (CDCl₃) δ 7.52–7.38 (m, 5H), 7.36–7.19 (m, 3H), 7.16–7.08 (m, 1H), 3.63 (s, 2H), 2.34 (s, 3H); ¹³C NMR (CDCl₃) δ 145.4, 141.9, 138.6, 138.1, 134.5, 128.8, 128.3, 127.4, 126.5, 123.7, 123.1, 118.7, 41.0, 15.5. Anal. Calcd for C₁₆H₁₄S: C, 80.63; H, 5.92; N, 0. Found: C, 80.94; H, 6.00; N, 0.

(±)-1-Methyl-2-(methylthio)-3-phenyl-1*H*-indene (13d). To a solution of 2-(1,2,3-benzotriazol-1-yl)-2-(methylthio)-1,1-diphenyl-1-butanole **11d** (0.5 g, 1.29 mmol) in 1,2-dichlor-ethane (20 mL) under nitrogen was added zinc bromide (0.87 g, 3.86 mmol), and the reaction mixture was refluxed for 5 min. The solvent was evaporated under reduced pressure, and the residue was purified by column chromatography on silica gel using hexanes to give **13d** as plates (78%): mp 73–74 °C; ¹H NMR (CDCl₃) δ 7.52–7.30 (m, 6H), 7.28–7.12 (m, 3H), 3.69 (q, J = 7.4 Hz, 1H), 2.24 (s, 3H), 1.46 (d, J = 7.4 Hz, 3H); ¹³C NMR (CDCl₃) δ 148.3, 144.0, 143.7, 140.0, 134.5, 129.2, 128.2, 127.5, 126.6, 124.6, 122.5, 119.4, 46.0, 17.6, 16.7. Anal. Calcd for C₁₇H₁₆S: C, 80.90; H, 6.39; N, 0. Found: C, 80.65; H, 6.61; N, 0.

(±)-3-Hydroxy-3-phenyl-2-butanone (14). To a solution of 3-(1,2,3-benzotriazol-1-yl)-3-(methylthio)-2-phenyl-2-butanol **11b** (0.3 g, 0.96 mmol) in THF (20 mL) under nitrogen was added zinc bromide (0.65 g, 2.87 mmol), and the reaction mixture was stirred for 3 h. The solvent was evaporated under reduced pressure, and the residue was purified by gradient column chromatography using the mixture of hexanes/EtOAc (9:1–3:1) to give **14** as colorless liquid (75%): ¹H NMR (CDCl₃) δ 7.45–7.27 (m, 5H), 4.59 (s, 1H), 2.07 (s, 3H), 1.76 (s, 3H); ¹³C NMR (CDCl₃) δ 209.6, 141.3, 128.6, 127.9, 125.8, 79.8, 24.0, 23.4. The spectral data of this compound is identical to that reported in the literature.¹⁷

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